

## Heterocyclic Thiones and Their Analogs in 1,3-Dipolar Cycloaddition Reactions: II.\* Reactions of Benzothiazole-2-thione with Nitrilimines

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**Abstract**—Benzothiazole-2-thione reacted with C,N-disubstituted nitrilimines to give the corresponding bis[2-(1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfides. In some cases,  $\alpha$ -(1,3-benzothiazol-2-ylsulfanyl)-( $\alpha$ -R)-alkanone arylhydrazones were formed as by-products. The structure of the isolated compounds was proved by X-ray analysis.

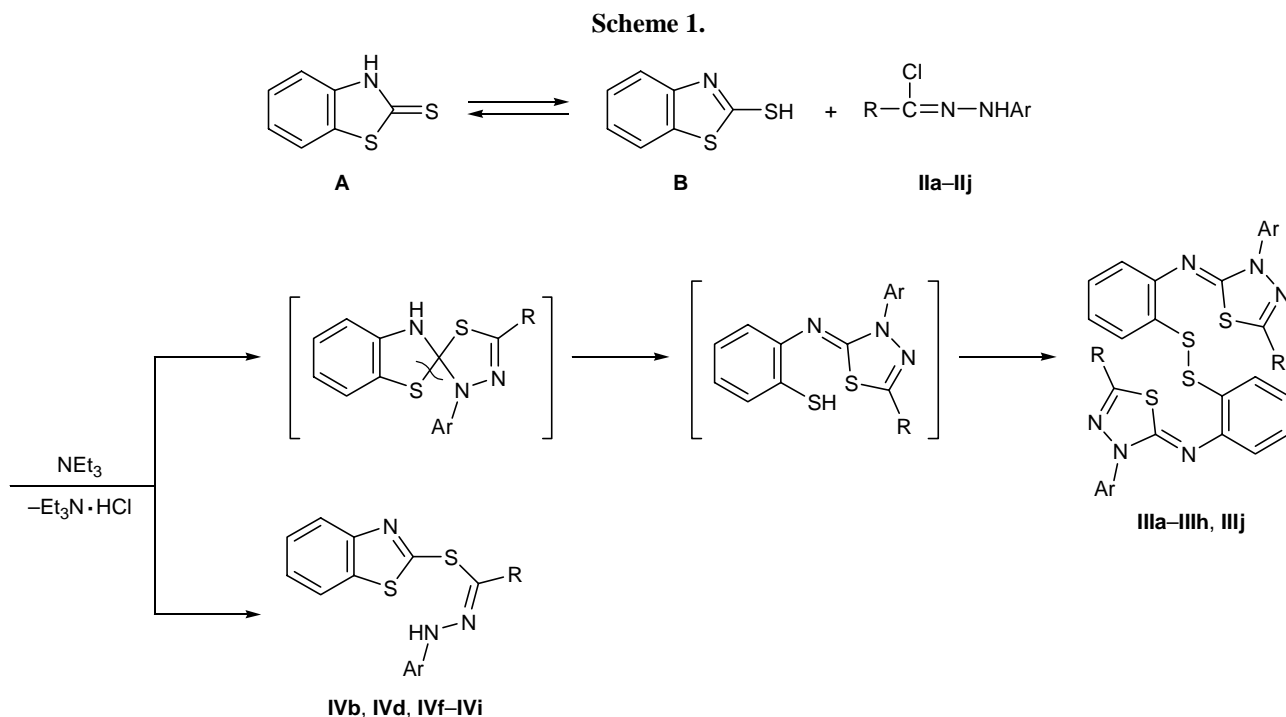
Reactions of thioketones with classical 1,3-dipolar systems have been well documented [2, 3]. Various acyclic and heterocyclic thioamides also react with 1,3-dipolar reagents [4–5]. We previously showed that 1,2-dithiophthalides containing an exocyclic thioketone moiety react with nitrilimines to give benzothiophene-spiro-thiadiazoles [1]. However, there are only fragmentary published data on the behavior in 1,3-dipolar cycloaddition reactions of compounds whose molecules contain a secondary amino group in the  $\alpha$ -position with respect to the thione group [7, 8].

The present article reports on the results of studying the reaction of benzothiazole-2-thione with C,N-disubstituted nitrilimines which were generated *in situ* from the corresponding *N*-aryl carbohydrazonoyl chlorides **IIa–IIj** by the action of triethylamine. Benzothiazole-2-thione possesses an ambident HN=C=S moiety which is capable of undergoing tautomeric transformation. Depending on the solvent polarity and basicity, benzothiazole-2-thione can exist as two tautomers, thione **A** or thiol **B** [9]; therefore, it could give rise to two series of derivatives with participation of the NH or SH functionality; furthermore, 1,3-dipolar cycloaddition involving the exocyclic thione group is possible [10–12].

Presumably, the initial stage in the reaction of benzothiazole-2-thione (**I**) with C,N-disubstituted nitrilimines is [3+2]-cyclization with participation of the thione C=S bond (tautomer **A**). The subsequent opening of the thiazole ring in the benzothiazole-spirothiadiazole structure thus formed, followed by dimerization of intermediate thiol, leads to formation of bis[2-(1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfides **IIIa–IIIh** and **IIIj** (Scheme 1). In some cases, we detected  $\alpha$ -(1,3-benzothiazol-2-ylsulfanyl)-( $\alpha$ -R)alkanone arylhydrazones **IVb**, **IVd**, and **IVf–IVi** which were formed as a result of nucleophilic substitution of the halogen atom in the carbohydrazonoyl chloride by the thiol tautomer (**B**) of initial benzothiazole-2-thione. In the reactions of benzothiazole-2-thione with *C*-ethoxycarbonyl-*N*-arylnitrilimines having an *ortho* substituent in the benzene ring, compounds **IVh** and **IVi** were the major products. It should be noted that hydrazones **IVb**, **IVd**, and **IVf–IVi** can exist as *E* and *Z* isomers. We succeeded in separating the *E* and *Z* isomers of arylhydrazone **IVh**, while compound **IVi** was isolated exclusively as the *E* isomer. In all other cases, mixtures of isomers were examined.

All the isolated products are crystalline substances which are stable on exposure to air for a long time. The <sup>1</sup>H NMR spectra of bis[2-(1,3,4-thiadiazole-2-ylideneamino)phenyl] disulfides **IIIa–IIIh** and **IIIj** contained

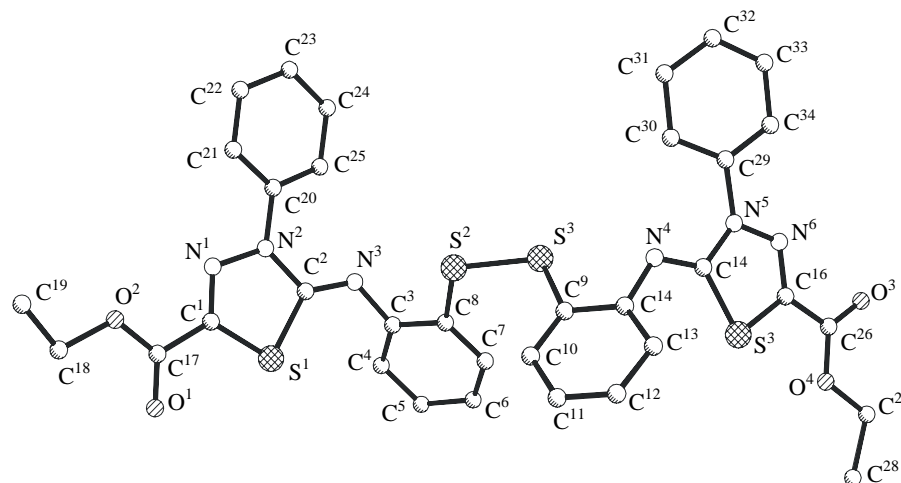
\* For communication I, see [1].



signals typical of substituents at the carbon and nitrogen atoms of the initial nitrilimine, as well as signals from aromatic protons. The <sup>13</sup>C NMR spectra of **IIIa-IIIh** and **IIIj** were consistent with the assumed structures. The formation of disulfides **III** was also confirmed by the mass spectra which contained the molecular ion peaks and peaks from ions formed by

cleavage of the S-S bond. Further fragmentation pattern is fairly complex.

The structure of bis[2-(5-ethoxycarbonyl-3-phenyl-2,3-dihydro-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (**IIIff**) was proved by X-ray analysis. The molecular structure is shown in Fig. 1 (hereinafter, the atom numbering given in the corresponding figure is



**Fig. 1.** Structure of the molecule of bis[2-(5-ethoxycarbonyl-3-phenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (**IIIff**) according to the X-ray diffraction data. Hydrogen atoms are not shown.

used while describing the X-ray diffraction data). A unit cell of **III**f includes two independent molecules (major *A* and minor *B*) which slightly differ from each other in mutual arrangement of the ethoxycarbonyl groups and benzene rings. A second-order pseudoaxis was revealed. It passes through points with approximate coordinates of [1/3, 1/3, 1/2]. The bond angles and bond lengths in the right part of the molecule slightly differ from those in the left part (Table 1). The torsion angles  $C^8S^2S^3C^9$  are 86.9 and 82.7° for molecules *A* and *B*, respectively. The thiadiazole rings are planar: the maximal deviations from the  $C^1N^1N^2C^2S^1$  plane are 0.03 and 0.06 Å, and from the  $C^{15}N^5N^6C^{16}S^4$  plane, 0.02 and 0.04 Å, respectively, for molecules *A* and *B*. However, the  $N^3$  and  $N^4$  atoms deviate from the thiadiazole ring planes; the deviations of  $N^3$  from the  $C^1N^1N^2C^2S^1$  plane are 0.15 and 0.13 Å, and of  $N^4$  from the  $C^{15}N^5N^6C^{16}S^4$  plane, 0.12 and 0.17 Å, respectively, for molecules *A* and *B*. Table 2 contains the dihedral angles between different planes in molecule **III**f.

As noted above,  $\alpha$ -(1,3-benzothiazol-2-ylsulfanyl)-( $\alpha$ -R)alkanone arylhydrazones **IV** exist as *E* and *Z* isomers. Apart from signals typical of substituents at the carbon and nitrogen atoms of the initial nitrilimine, the  $^1H$  NMR spectra of compounds **IV** contained a singlet from proton of the hydrazone moiety at  $\delta$  12.95–14.05 (*E* isomers) and 9.80–12.9 ppm (*Z* isomers). The  $^{13}C$  NMR spectra of **IV**b, **IV**d, and **IV**f–**IV**i were consistent with the assumed structures. Compounds **IV**b, **IV**d, and **IV**f–**IV**i showed in the mass spectra the corresponding molecular ion peaks and peaks from ions formed by cleavage of the thiol C–S bond. As with compounds **III**, further fragmentation pattern of **IV** was fairly complex.

The structure of compounds **IV**f and **IV**h was proved by X-ray analysis (Figs. 2, 3). The X-ray diffraction data for compound **IV**f showed formation of intramolecular hydrogen bond  $O^1 \cdots H(N^3)$  which stabilizes the *E* isomer. The hydrogen bond parameters are as follows:  $N^3-H$  0.841,  $O^1 \cdots H(N^3)$  1.967,  $O^1 \cdots N^3$

**Table 1.** Bond lengths (*d*, Å) and bond angles ( $\omega$ , deg) in molecules *A* and *B* of bis[2-(5-ethoxycarbonyl-3-phenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (**III**f)<sup>a</sup>

Bond	<i>d</i>		Bond	<i>d</i>	
	molecule <i>A</i>	molecule <i>B</i>		molecule <i>A</i>	molecule <i>B</i>
S <sup>1</sup> –C <sup>1</sup>	1.722(5)	1.738(5)	N <sup>5</sup> –N <sup>6</sup>	1.359(5)	1.367(5)
S <sup>1</sup> –C <sup>2</sup>	1.771(5)	1.764(5)	N <sup>5</sup> –C <sup>15</sup>	1.397(5)	1.393(5)
S <sup>2</sup> –C <sup>8</sup>	1.783(4)	1.783(4)	N <sup>5</sup> –C <sup>29</sup>	1.444(6)	1.434(6)
S <sup>2</sup> –S <sup>3</sup>	2.0314(16)	2.0296(15)	N <sup>6</sup> –C <sup>16</sup>	1.282(6)	1.279(6)
S <sup>3</sup> –C <sup>9</sup>	1.785(4)	1.787(4)	C <sup>1</sup> –C <sup>17</sup>	1.487(7)	1.486(7)
S <sup>4</sup> –C <sup>16</sup>	1.734(5)	1.732(5)	C <sup>3</sup> –C <sup>4</sup>	1.390(6)	1.391(6)
S <sup>4</sup> –C <sup>15</sup>	1.778(5)	1.771(4)	C <sup>3</sup> –C <sup>8</sup>	1.393(6)	1.393(6)
O <sup>1</sup> –C <sup>17</sup>	1.195(6)	1.191(6)	C <sup>4</sup> –C <sup>5</sup>	1.385(7)	1.390(7)
O <sup>2</sup> –C <sup>17</sup>	1.311(6)	1.336(7)	C <sup>5</sup> –C <sup>6</sup>	1.372(7)	1.374(7)
O <sup>2</sup> –C <sup>18</sup>	1.460(6)	1.475(6)	C <sup>6</sup> –C <sup>7</sup>	1.389(6)	1.387(6)
O <sup>3</sup> –C <sup>26</sup>	1.194(6)	1.197(6)	C <sup>7</sup> –C <sup>8</sup>	1.382(6)	1.385(6)
O <sup>4</sup> –C <sup>26</sup>	1.323(6)	1.317(6)	C <sup>9</sup> –C <sup>10</sup>	1.381(6)	1.381(5)
O <sup>4</sup> –C <sup>27</sup>	1.461(5)	1.468(6)	C <sup>9</sup> –C <sup>14</sup>	1.392(6)	1.397(5)
N <sup>1</sup> –C <sup>1</sup>	1.278(6)	1.276(6)	C <sup>10</sup> –C <sup>11</sup>	1.397(7)	1.393(6)
N <sup>1</sup> –N <sup>2</sup>	1.369(5)	1.370(5)	C <sup>11</sup> –C <sup>12</sup>	1.363(7)	1.371(7)
N <sup>2</sup> –C <sup>2</sup>	1.398(5)	1.412(6)	C <sup>12</sup> –C <sup>13</sup>	1.393(7)	1.390(6)
N <sup>2</sup> –C <sup>20</sup>	1.437(5)	1.434(6)	C <sup>13</sup> –C <sup>14</sup>	1.394(6)	1.396(6)
N <sup>3</sup> –C <sup>2</sup>	1.257(5)	1.259(5)	C <sup>16</sup> –C <sup>26</sup>	1.489(7)	1.486(7)
N <sup>3</sup> –C <sup>3</sup>	1.415(5)	1.417(5)	C <sup>18</sup> –C <sup>19</sup>	1.435(8)	1.480(8)
N <sup>4</sup> –C <sup>15</sup>	1.266(5)	1.265(5)	C <sup>27</sup> –C <sup>28</sup>	1.493(8)	1.409(8)
N <sup>4</sup> –C <sup>14</sup>	1.411(5)	1.408(5)			

Table 1. (Contd.)

Angle	$\omega$		Angle	$\omega$	
	molecule A	molecule B		molecule A	molecule B
C <sup>1</sup> S <sup>1</sup> C <sup>2</sup>	88.4(2)	88.8(3)	C <sup>7</sup> C <sup>8</sup> S <sup>2</sup>	124.6(4)	125.1(3)
C <sup>8</sup> S <sup>2</sup> S <sup>3</sup>	105.02(15)	106.45(15)	C <sup>3</sup> C <sup>8</sup> S <sup>2</sup>	114.9(3)	114.4(3)
C <sup>9</sup> S <sup>3</sup> S <sup>2</sup>	106.48(15)	105.12(14)	C <sup>10</sup> C <sup>9</sup> C <sup>14</sup>	120.7(4)	121.1(4)
C <sup>16</sup> S <sup>4</sup> C <sup>15</sup>	88.4(2)	87.7(2)	C <sup>10</sup> C <sup>9</sup> S <sup>3</sup>	124.9(3)	124.9(3)
C <sup>17</sup> O <sup>2</sup> C <sup>18</sup>	116.1(4)	117.4(5)	C <sup>14</sup> C <sup>9</sup> S <sup>3</sup>	114.4(3)	114.0(3)
C <sup>26</sup> O <sup>4</sup> C <sup>27</sup>	116.4(4)	116.1(4)	C <sup>9</sup> C <sup>10</sup> C <sup>11</sup>	118.8(4)	119.2(4)
C <sup>1</sup> N <sup>1</sup> N <sup>2</sup>	110.9(4)	111.5(4)	C <sup>12</sup> C <sup>11</sup> C <sup>10</sup>	121.2(4)	120.1(4)
N <sup>1</sup> N <sup>2</sup> C <sup>2</sup>	115.8(4)	115.6(4)	C <sup>11</sup> C <sup>12</sup> C <sup>13</sup>	120.2(5)	121.2(4)
N <sup>1</sup> N <sup>2</sup> C <sup>20</sup>	116.6(3)	117.0(4)	C <sup>14</sup> C <sup>13</sup> C <sup>12</sup>	119.5(4)	119.3(4)
C <sup>2</sup> N <sup>2</sup> C <sup>20</sup>	127.2(4)	127.1(4)	C <sup>13</sup> C <sup>14</sup> C <sup>9</sup>	119.6(4)	119.1(4)
C <sup>2</sup> -N <sup>3</sup> -C <sup>3</sup>	122.7(4)	122.5(4)	C <sup>13</sup> C <sup>14</sup> N <sup>4</sup>	122.9(4)	123.8(4)
C <sup>15</sup> N <sup>4</sup> C <sup>14</sup>	121.0(4)	122.4(4)	C <sup>9</sup> C <sup>14</sup> N <sup>4</sup>	117.3(4)	116.9(4)
N <sup>6</sup> N <sup>5</sup> C <sup>15</sup>	116.7(4)	115.5(4)	N <sup>4</sup> C <sup>15</sup> N <sup>5</sup>	124.6(4)	123.7(4)
N <sup>6</sup> N <sup>5</sup> C <sup>29</sup>	116.5(4)	116.5(4)	N <sup>4</sup> C <sup>15</sup> S <sup>4</sup>	128.3(3)	127.9(3)
C <sup>15</sup> N <sup>5</sup> C <sup>29</sup>	126.8(4)	127.6(4)	N <sup>5</sup> C <sup>15</sup> S <sup>4</sup>	107.1(3)	108.3(3)
C <sup>16</sup> N <sup>6</sup> N <sup>5</sup>	111.1(4)	111.1(4)	N <sup>6</sup> C <sup>16</sup> C <sup>26</sup>	120.9(5)	123.2(5)
N <sup>1</sup> C <sup>1</sup> C <sup>17</sup>	123.1(5)	120.8(5)	N <sup>6</sup> C <sup>16</sup> S <sup>4</sup>	116.6(3)	117.1(4)
N <sup>1</sup> C <sup>1</sup> S <sup>1</sup>	117.2(4)	116.5(4)	C <sup>26</sup> C <sup>16</sup> S <sup>4</sup>	122.5(4)	119.7(4)
C <sup>17</sup> C <sup>1</sup> S <sup>1</sup>	119.7(4)	122.7(5)	O <sup>1</sup> C <sup>17</sup> O <sup>2</sup>	126.4(5)	125.6(5)
N <sup>3</sup> C <sup>2</sup> N <sup>2</sup>	124.3(4)	122.6(4)	O <sup>1</sup> C <sup>17</sup> C <sup>1</sup>	121.9(5)	125.2(7)
N <sup>3</sup> C <sup>2</sup> S <sup>1</sup>	128.1(4)	129.8(4)	O <sup>2</sup> C <sup>17</sup> C <sup>1</sup>	111.7(5)	109.2(5)
N <sup>2</sup> C <sup>2</sup> S <sup>1</sup>	107.5(3)	107.5(3)	C <sup>19</sup> C <sup>18</sup> O <sup>2</sup>	107.5(5)	107.2(5)
C <sup>4</sup> C <sup>3</sup> C <sup>8</sup>	119.5(4)	119.8(4)	C <sup>25</sup> C <sup>20</sup> N <sup>2</sup>	121.3(4)	119.2(5)
C <sup>4</sup> C <sup>3</sup> N <sup>3</sup>	123.5(4)	123.2(4)	C <sup>21</sup> C <sup>20</sup> N <sup>2</sup>	118.1(4)	121.9(5)
C <sup>8</sup> C <sup>3</sup> N <sup>3</sup>	116.7(4)	116.7(4)	O <sup>3</sup> C <sup>26</sup> O <sup>4</sup>	125.3(5)	126.3(5)
C <sup>5</sup> C <sup>4</sup> C <sup>3</sup>	119.4(5)	119.2(4)	O <sup>3</sup> C <sup>26</sup> C <sup>16</sup>	124.6(5)	122.5(5)
C <sup>6</sup> C <sup>5</sup> C <sup>4</sup>	121.1(5)	120.9(5)	O <sup>4</sup> C <sup>26</sup> C <sup>16</sup>	110.1(5)	111.2(5)
C <sup>5</sup> C <sup>6</sup> C <sup>7</sup>	119.8(4)	120.1(5)	O <sup>4</sup> C <sup>27</sup> C <sup>28</sup>	107.1(4)	107.7(5)
C <sup>8</sup> C <sup>7</sup> C <sup>6</sup>	119.6(4)	119.5(4)	C <sup>30</sup> C <sup>29</sup> N <sup>5</sup>	121.2(4)	118.2(5)
C <sup>7</sup> C <sup>8</sup> C <sup>3</sup>	120.5(4)	120.4(4)	C <sup>34</sup> C <sup>29</sup> N <sup>5</sup>	118.0(4)	120.8(4)

<sup>a</sup> Hereinafter, the bond lengths and bond angles in the benzene rings are not given, for they have their standard values.

2.622 Å;  $\angle$ N<sup>3</sup>HO<sup>1</sup> 133.9°. The benzothiazole ring is planar: the maximal deviation of atoms from the mean-square plane is 0.008 Å; the maximal deviation of atoms from the HN<sup>3</sup>N<sup>2</sup>C<sup>14</sup>C<sup>15</sup>O<sup>1</sup> plane arising from hydrogen bonding is 0.03 Å. The S<sup>2</sup> atom appears in both the benzothiazole ring and HN<sup>3</sup>N<sup>2</sup>C<sup>14</sup>C<sup>15</sup>O<sup>1</sup> planes. The existence of conjugation between the C<sup>14</sup>=N<sup>2</sup> double bond, lone electron pair on the N<sup>3</sup> atom, and C<sup>8</sup>-C<sup>13</sup> benzene ring may be presumed: the angle between the benzene ring and HN<sup>3</sup>N<sup>2</sup>C<sup>14</sup>C<sup>15</sup>O<sup>1</sup>

planes is as small as 6.8°. The bond lengths and bond angles in molecule **IVf** are listed in Table 3.

Ethyl 2-(1,3-benzothiazol-2-ylsulfanyl)-2-(*o*-nitrophenylhydrazono)acetate (**IVh**) in crystal also has *E* configuration of the hydrazone fragment (Fig. 3), which is stabilized by intramolecular hydrogen bond O<sup>1</sup>...H(N<sup>3</sup>) and weak intramolecular hydrogen bond O<sup>4</sup>...H(N<sup>3</sup>) with the following parameters: O<sup>1</sup>...H(N<sup>3</sup>): N<sup>3</sup>-H 0.820, O<sup>1</sup>-H(N<sup>3</sup>) 2.11, O<sup>1</sup>-N<sup>3</sup> 2.640 Å;  $\angle$ N<sup>3</sup>HO<sup>1</sup> 122.2°; O<sup>4</sup>...H(N<sup>3</sup>): N<sup>3</sup>-H 0.820, O<sup>4</sup>-H(N<sup>3</sup>) 1.94,

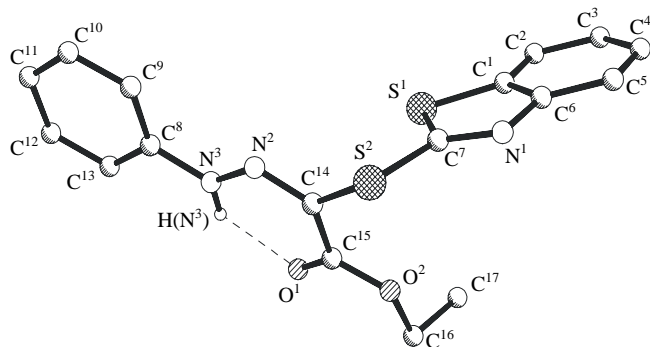
**Table 2.** Dihedral angles between planes *x* and *y* in molecules *A* and *B*<sup>a</sup> of bis[2-(5-ethoxycarbonyl-3-phenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (**III**f)

Plane <i>x</i>	Plane <i>y</i>		
	C <sup>1</sup> N <sup>1</sup> N <sup>2</sup> C <sup>2</sup> S <sup>1</sup>	C <sup>15</sup> N <sup>3</sup> N <sup>6</sup> C <sup>16</sup> S <sup>4</sup>	C <sup>3</sup> -C <sup>8</sup>
O <sup>1</sup> C <sup>17</sup> O <sup>2</sup>	5.8 (12.2)	–	–
C <sup>20</sup> -C <sup>25</sup>	12.9 (11.1)	–	–
C <sup>3</sup> -C <sup>8</sup>	57.2 (54.6)	–	–
O <sup>3</sup> C <sup>26</sup> O <sup>4</sup>	–	4.3 (12.7)	–
C <sup>29</sup> -C <sup>34</sup>	–	23.1 (11.6)	–
C <sup>9</sup> -C <sup>14</sup>	–	56.4 (58.1)	77.2 (71.9)

<sup>a</sup> In parentheses.

O<sup>4</sup>-N<sup>3</sup> 2.620 Å; ∠N<sup>3</sup>HO<sup>4</sup> 138.7°. The maximal deviations of atoms from the HN<sup>3</sup>N<sup>2</sup>C<sup>8</sup>C<sup>9</sup>O<sup>1</sup> plane formed as a result of H-bonding O<sup>1</sup>⋯H(N<sup>3</sup>) is 0.04 Å. The HN<sup>3</sup>C<sup>12</sup>C<sup>13</sup>N<sup>4</sup>O<sup>4</sup> ring closed via O<sup>4</sup>⋯H(N<sup>3</sup>) hydrogen bond is planar within 0.005 Å. The benzothiazole ring S<sup>1</sup>C<sup>2</sup>C<sup>7</sup>N<sup>1</sup>C<sup>1</sup> is also planar: the maximal deviation of atoms from that plane is 0.01 Å. The S<sup>2</sup> atom simultaneously lies in the plane of the benzothiazole ring and in the HN<sup>3</sup>N<sup>2</sup>C<sup>8</sup>C<sup>9</sup>O<sup>1</sup> plane. The C<sup>8</sup>=N<sup>2</sup> double bond, lone electron pair on the N<sup>3</sup> atom, and the C<sup>12</sup>-C<sup>17</sup> benzene ring are likely to be involved in conjugation: the dihedral angle between the benzene ring plane and HN<sup>3</sup>C<sup>12</sup>C<sup>13</sup>O<sup>1</sup>N<sup>4</sup>O<sup>4</sup> H-bond plane is 0.8°. The nitro group N<sup>4</sup>O<sup>4</sup>O<sup>3</sup> is conjugated with the benzene ring: the dihedral angle between the respective planes is 0.6°. The bond angles and bond lengths in molecule **IV**h are given in Table 4.

Thus we have found that the reactions of benzothiazole-2-thione with equimolar amounts of C,N-disubstituted nitrilimines begin with 1,3-dipolar cycloaddition at the exocyclic C=S bond to give an unstable

**Fig. 2.** Structure of the molecule of ethyl (1,3-benzothiazol-2-ylsulfanyl)(phenylhydrazono)acetate (**IV**f) according to the X-ray diffraction data. Only the hydrogen atom involved in intramolecular hydrogen bond is shown.

spiro compound which readily decomposes via cleavage of the C-S bond in the thiazole ring. The subsequent dimerization of intermediate thiol results in formation of substituted bis[2-(1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide **III**. Compounds formed via replacement of the halogen atom in carbonyl chlorides by the thiol group of benzothiazole-2-thione tautomer, α-(1,3-benzothiazol-2-ylsulfanyl)-(α-R)alkanone arylhydrazones **IV**, are as a rule, minor products.

## EXPERIMENTAL

The <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Bruker AM-500 spectrometer (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C) from 20% solutions in DMSO-*d*<sub>6</sub>. The mass spectra (electron impact, 70 eV) were obtained on an MKh-1321 instrument (vaporizer temperature 120°C, ion source temperature 200°C).

X-Ray analysis of single crystals of bis[2-(5-ethoxycarbonyl-3-phenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (**III**f) was performed on a Syntex P-1 automatic diffractometer (CuK<sub>α</sub> irradiation, β-filter, θ/2θ scanning in the range 2.30 ≤ Θ ≤ 57.44°). Monoclinic crystals, C<sub>34</sub>H<sub>28</sub>N<sub>6</sub>O<sub>4</sub>S<sub>4</sub>, with the following unit cell parameters: *a* = 9.0270(10), *b* = 30.925(4), *c* = 24.575(4) Å; β = 95.630(10)°; *V* = 6827.3(16) Å<sup>3</sup>; space group *P*2<sub>1</sub>/*c*; *Z* = 8, *d*<sub>calc</sub> = 1.387 g/cm<sup>3</sup>. The structure was solved by the direct method using 7094 reflections with σ > 3σ(*I*) in full-matrix anisotropic approximation for non-hydrogen atoms and isotropic approximation for hydrogen atoms; the final divergence factors were *R* = 0.0497 and *R*<sub>w</sub> = 0.1084 (no correction for absorption was introduced).

X-Ray diffraction data for ethyl 2-(1,3-benzothiazol-2-ylsulfanyl)-2-(phenylhydrazono)acetate (**IV**f) were obtained on a CAD-4 automatic diffractometer (MoK<sub>α</sub> irradiation, β-filter, θ/2θ scanning in the range 2.08 ≤ Θ ≤ 24.97°). Triclinic crystals, C<sub>17</sub>H<sub>15</sub>N<sub>3</sub>O<sub>2</sub>S<sub>2</sub>, with the following unit cell parameters: *a* = 9.311(1), *b* = 9.540(1), *c* = 10.179(2) Å; α = 104.13(2), β = 95.74(2), γ = 94.84(2)°; *V* = 866.9(2) Å<sup>3</sup>; space group *P*1; *Z* = 2; *d*<sub>calc</sub> = 1.369 g/cm<sup>3</sup>. The structure was solved by the direct method using 3330 reflections with σ > 3σ(*I*) in full-matrix anisotropic approximation for non-hydrogen atoms and isotropic approximation for hydrogen atoms; the final divergence factors were *R* = 0.0254 and *R*<sub>w</sub> = 0.0692 (no correction for absorption was introduced).

X-Ray diffraction data for ethyl 2-(1,3-benzothiazol-2-ylsulfanyl)-2-(*o*-nitrophenylhydrazono)-acetate (**IVh**) were obtained on an Enraf–Nonius CAD-4 automatic diffractometer (MoK $\alpha$  irradiation,  $\beta$ -filter,  $\theta/2\theta$  scanning in the range  $2.48 \leq \Theta \leq 24.91^\circ$ ). Monoclinic crystals, C<sub>17</sub>H<sub>14</sub>N<sub>4</sub>O<sub>4</sub>S<sub>2</sub>, with the following unit cell parameters:  $a = 7.1060(10)$ ,  $b = 29.722(6)$ ,  $c = 8.5500(10)$  Å;  $\beta = 93.03(3)^\circ$ ;  $V = 1803.3(5)$  Å<sup>3</sup>; space group  $P2_1/c$ ;  $Z = 4$ ;  $d_{\text{calc}} = 1.482$  g/cm<sup>3</sup>. The structure was solved by the direct method using 1200 reflections with  $\sigma > 3\sigma(I)$  in full-matrix anisotropic approximation for non-hydrogen atoms and isotropic approximation for hydrogen atoms; the final divergence factors were  $R = 0.0806$  and  $R_w = 0.1784$  (no correction for absorption was introduced).

The complete sets of crystallographic coordinates of non-hydrogen and hydrogen atoms and their equivalent temperature factors are available from the authors.

Nitrilimines were generated *in situ* by the action of triethylamine on the corresponding *N*-aryl carbohydrazonoyl chlorides **IIa–IIj** [13].

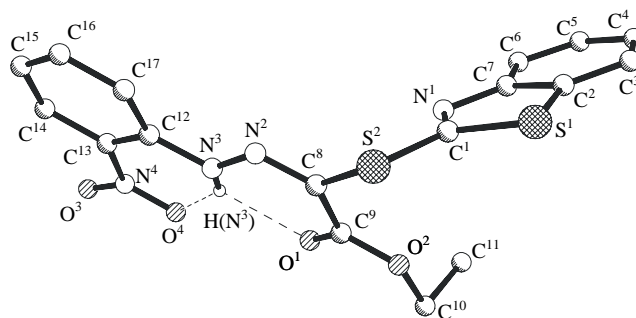
**General procedure for reactions of benzothiazole-2-thione with C-aryl(acetyl, methoxycarbonyl, ethoxycarbonyl)-N-phenylnitrilimines.** To a solution of 4 mmol of benzothiazole-2-thione (**I**) in 50 ml of anhydrous toluene we added in succession 4 mmol of the corresponding *N*-aryl carbohydrazonoyl chloride **II** and 4.5 mmol of anhydrous triethylamine. The mixture was heated for 4 h under reflux, cooled, and filtered from triethylamine hydrochloride (yield 75–90%). The filtrate was evaporated under reduced pressure, and the oily residue was crystallized by grinding with diethyl ether. Products **III** and **IV** were separated by column chromatography on silica gel using chloroform as eluent. Compound **IVh** was separated into the *E* and *Z* isomers, and only the *E* isomer of **IVi** was isolated from the mixture; in the other cases, *E/Z* isomer mixtures were analyzed. The products were additionally recrystallized from acetone.

**Bis[2-(5-acetyl-3-phenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (IIIa).** Yield 75%, mp 210–211°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.59 s (3H, CH<sub>3</sub>CO), 7.1–7.8 m (18H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum,  $\delta_c$ , ppm: 156.71 (C<sup>2</sup>); 148.27 (C<sup>5</sup>); 189.23 (CH<sub>3</sub>CO); 24.86 (CH<sub>3</sub>CO); 123.52, 128.41, 128.97, 138.05 (NC<sub>6</sub>H<sub>5</sub>); 118.23, 125.27, 126.86, 127.85, 128.68, 147.15 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 652 (40) [ $M$ ]<sup>+</sup>, 326 (50) [C<sub>16</sub>H<sub>12</sub>N<sub>3</sub>OS<sub>2</sub>]<sup>+</sup>, 294 (10) [C<sub>16</sub>H<sub>12</sub>N<sub>3</sub>OS]<sup>+</sup>, 257 (40) [C<sub>13</sub>H<sub>9</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 225 (15)

**Table 3.** Bond lengths ( $d$ , Å) and bond angles ( $\omega$ , deg) in the molecule of ethyl (1,3-benzothiazol-2-ylsulfanyl)-(phenylhydrazono)acetate (**IVf**)

Bond	$d$	Bond	$d$
S <sup>1</sup> –C <sup>1</sup>	1.7337(19)	N <sup>2</sup> –N <sup>3</sup>	1.314(2)
S <sup>1</sup> –C <sup>7</sup>	1.7411(19)	N <sup>3</sup> –C <sup>8</sup>	1.399(3)
S <sup>2</sup> –C <sup>7</sup>	1.749(2)	C <sup>1</sup> –C <sup>2</sup>	1.390(3)
S <sup>2</sup> –C <sup>14</sup>	1.7662(19)	C <sup>1</sup> –C <sup>6</sup>	1.402(3)
O <sup>1</sup> –C <sup>15</sup>	1.212(2)	C <sup>2</sup> –C <sup>3</sup>	1.376(3)
O <sup>2</sup> –C <sup>15</sup>	1.324(2)	C <sup>3</sup> –C <sup>4</sup>	1.391(4)
O <sup>2</sup> –C <sup>16</sup>	1.461(3)	C <sup>4</sup> –C <sup>5</sup>	1.379(3)
N <sup>1</sup> –C <sup>7</sup>	1.288(2)	C <sup>5</sup> –C <sup>6</sup>	1.389(3)
N <sup>1</sup> –C <sup>6</sup>	1.393(2)	C <sup>14</sup> –C <sup>15</sup>	1.479(3)
N <sup>2</sup> –C <sup>14</sup>	1.300(2)	C <sup>16</sup> –C <sup>17</sup>	1.457(4)
Angle	$\omega$	Angle	$\omega$
C <sup>1</sup> S <sup>1</sup> C <sup>7</sup>	88.17(9)	C <sup>5</sup> C <sup>6</sup> C <sup>1</sup>	119.8(2)
C <sup>7</sup> S <sup>2</sup> C <sup>14</sup>	100.27(9)	N <sup>1</sup> C <sup>6</sup> C <sup>1</sup>	115.12(17)
C <sup>15</sup> O <sup>2</sup> C <sup>16</sup>	117.46(17)	N <sup>1</sup> C <sup>7</sup> S <sup>1</sup>	117.63(14)
C <sup>7</sup> N <sup>1</sup> C <sup>6</sup>	109.45(16)	N <sup>1</sup> C <sup>7</sup> S <sup>2</sup>	121.20(14)
C <sup>14</sup> N <sup>2</sup> N <sup>3</sup>	120.61(16)	S <sup>1</sup> C <sup>7</sup> S <sup>2</sup>	121.17(11)
N <sup>2</sup> N <sup>3</sup> C <sup>8</sup>	120.94(17)	C <sup>9</sup> C <sup>8</sup> N <sup>3</sup>	122.15(19)
C <sup>2</sup> C <sup>1</sup> C <sup>6</sup>	121.48(19)	C <sup>13</sup> C <sup>8</sup> N <sup>3</sup>	117.84(18)
C <sup>2</sup> C <sup>1</sup> S <sup>1</sup>	128.90(16)	N <sup>2</sup> C <sup>14</sup> C <sup>15</sup>	124.59(17)
C <sup>6</sup> C <sup>1</sup> S <sup>1</sup>	109.62(14)	N <sup>2</sup> C <sup>14</sup> S <sup>2</sup>	114.75(14)
C <sup>3</sup> C <sup>2</sup> C <sup>1</sup>	117.8(2)	C <sup>15</sup> C <sup>14</sup> S <sup>2</sup>	120.60(15)
C <sup>2</sup> C <sup>3</sup> C <sup>4</sup>	121.1(2)	O <sup>1</sup> C <sup>15</sup> O <sup>2</sup>	124.57(18)
C <sup>5</sup> C <sup>4</sup> C <sup>3</sup>	121.3(2)	O <sup>1</sup> C <sup>15</sup> C <sup>14</sup>	121.94(18)
C <sup>4</sup> C <sup>5</sup> C <sup>6</sup>	118.5(2)	O <sup>2</sup> C <sup>15</sup> C <sup>14</sup>	113.49(16)
C <sup>5</sup> C <sup>6</sup> N <sup>1</sup>	125.09(19)	C <sup>17</sup> C <sup>16</sup> O <sup>2</sup>	109.6(2)

[C<sub>8</sub>H<sub>5</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 167 (40) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (80) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 108 (20) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 105 (40) [C<sub>6</sub>H<sub>5</sub>N<sub>2</sub>]<sup>+</sup>, 91 (70) [C<sub>6</sub>H<sub>5</sub>N]<sup>+</sup>. Found, %: C 58.51; H 3.71; N 12.69;



**Fig. 3.** Structure of the molecule of ethyl (1,3-benzothiazol-2-ylsulfanyl)-(o-nitrophenylhydrazono)acetate (**IVh**) according to the X-ray diffraction data. Only the hydrogen atom involved in intramolecular hydrogen bonds is shown.

**Table 4.** Bond lengths (*d*, Å) and bond angles ( $\omega$ , deg) in the molecule of ethyl (1,3-benzothiazol-2-ylsulfanyl)-(*o*-nitrophenylhydrazono)acetate (**IVh**)

Bond	<i>d</i>	Bond	<i>d</i>
S <sup>1</sup> –C <sup>2</sup>	1.728(11)	N <sup>3</sup> –C <sup>12</sup>	1.382(13)
S <sup>1</sup> –C <sup>1</sup>	1.737(10)	N <sup>4</sup> –C <sup>13</sup>	1.467(16)
S <sup>2</sup> –C <sup>1</sup>	1.747(10)	N <sup>2</sup> –N <sup>3</sup>	1.318(11)
S <sup>2</sup> –C <sup>8</sup>	1.761(10)	C <sup>2</sup> –C <sup>3</sup>	1.383(15)
O <sup>1</sup> –C <sup>9</sup>	1.208(12)	C <sup>2</sup> –C <sup>7</sup>	1.406(14)
O <sup>2</sup> –C <sup>9</sup>	1.307(13)	C <sup>3</sup> –C <sup>4</sup>	1.381(19)
O <sup>2</sup> –C <sup>10</sup>	1.462(16)	C <sup>4</sup> –C <sup>5</sup>	1.35(2)
O <sup>3</sup> –N <sup>4</sup>	1.198(12)	C <sup>5</sup> –C <sup>6</sup>	1.354(18)
O <sup>4</sup> –N <sup>4</sup>	1.214(12)	C <sup>6</sup> –C <sup>7</sup>	1.386(16)
N <sup>1</sup> –C <sup>1</sup>	1.285(12)	C <sup>8</sup> –C <sup>9</sup>	1.499(15)
N <sup>1</sup> –C <sup>7</sup>	1.383(12)	C <sup>10</sup> –C <sup>11</sup>	1.48(2)
N <sup>2</sup> –C <sup>8</sup>	1.293(12)		
Angle	$\omega$	Angle	$\omega$
C <sup>2</sup> S <sup>1</sup> C <sup>1</sup>	89.1(5)	C <sup>5</sup> C <sup>4</sup> C <sup>3</sup>	121.2(13)
C <sup>1</sup> S <sup>2</sup> C <sup>8</sup>	100.0(5)	C <sup>4</sup> C <sup>5</sup> C <sup>6</sup>	121.9(14)
C <sup>9</sup> O <sup>2</sup> C <sup>10</sup>	116.0(10)	C <sup>5</sup> C <sup>6</sup> C <sup>7</sup>	119.6(13)
C <sup>1</sup> N <sup>1</sup> C <sup>7</sup>	110.5(8)	N <sup>1</sup> C <sup>7</sup> C <sup>2</sup>	115.1(9)
C <sup>8</sup> N <sup>2</sup> N <sup>3</sup>	120.1(9)	N <sup>1</sup> C <sup>7</sup> C <sup>6</sup>	126.5(11)
N <sup>2</sup> N <sup>3</sup> C <sup>12</sup>	117.5(9)	C <sup>2</sup> C <sup>7</sup> C <sup>6</sup>	118.4(11)
O <sup>3</sup> N <sup>4</sup> O <sup>4</sup>	122.3(12)	N <sup>2</sup> C <sup>8</sup> C <sup>9</sup>	124.6(9)
O <sup>3</sup> N <sup>4</sup> C <sup>13</sup>	119.3(12)	N <sup>2</sup> C <sup>8</sup> S <sup>2</sup>	115.7(8)
O <sup>4</sup> N <sup>4</sup> C <sup>13</sup>	118.3(10)	C <sup>9</sup> C <sup>8</sup> S <sup>2</sup>	119.7(8)
N <sup>1</sup> C <sup>1</sup> S <sup>2</sup>	125.3(7)	O <sup>1</sup> C <sup>9</sup> O <sup>2</sup>	125.2(10)
N <sup>1</sup> C <sup>1</sup> S <sup>1</sup>	116.4(7)	O <sup>1</sup> C <sup>9</sup> C <sup>8</sup>	122.4(10)
S <sup>2</sup> C <sup>1</sup> S <sup>1</sup>	118.2(6)	O <sup>2</sup> C <sup>9</sup> C <sup>8</sup>	112.4(9)
C <sup>3</sup> C <sup>2</sup> C <sup>7</sup>	121.0(11)	C <sup>11</sup> C <sup>10</sup> O <sup>2</sup>	108.4(13)
C <sup>3</sup> C <sup>2</sup> S <sup>1</sup>	130.1(9)	N <sup>3</sup> C <sup>12</sup> C <sup>13</sup>	121.9(11)
C <sup>7</sup> C <sup>2</sup> S <sup>1</sup>	108.8(7)	N <sup>3</sup> C <sup>12</sup> C <sup>17</sup>	121.6(10)
C <sup>4</sup> C <sup>3</sup> C <sup>2</sup>	117.8(13)		

S 19.11. C<sub>32</sub>H<sub>24</sub>N<sub>6</sub>O<sub>2</sub>S<sub>4</sub>. Calculated, %: C 58.82; H 3.68; N 12.87; S 19.60.

**Bis[2-(5-acetyl-3-*m*-chlorophenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (IIIb).** Yield 59%, mp 207–208°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.6 s (3H, CH<sub>3</sub>CO), 7.1–8.2 m (16H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 148.02 (C<sup>2</sup>); 147.52 (C<sup>5</sup>); 189.29 (CH<sub>3</sub>CO); 24.93 (CH<sub>3</sub>CO); 121.62, 122.81, 128.51, 128.81, 133.16 (NC<sub>6</sub>H<sub>4</sub>Cl-*m*); 118.10, 125.46, 127.03, 127.48, 130.64, 139.03 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 720 (10) [M]<sup>+</sup>, 360 (50) [C<sub>16</sub>H<sub>11</sub>CIN<sub>3</sub>OS<sub>2</sub>]<sup>+</sup>, 328 (10) [C<sub>16</sub>H<sub>11</sub>C<sub>6</sub>N<sub>3</sub>OS]<sup>+</sup>, 256 (50) [C<sub>13</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>,

224 (70) [C<sub>8</sub>H<sub>4</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 167 (60) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (90) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 139 (40) [C<sub>6</sub>H<sub>4</sub>CIN<sub>2</sub>]<sup>+</sup>, 108 (20) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (30) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>. Found, %: C 53.64; H 3.60; N 10.99; S 17.77. C<sub>32</sub>H<sub>22</sub>Cl<sub>2</sub>N<sub>6</sub>O<sub>2</sub>S<sub>4</sub>. Calculated, %: C 53.21; H 3.05; N 11.64; S 17.74.

**Bis[2-(5-acetyl-3-*o*-chlorophenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (IIIc).** Yield 74%, mp 127–128°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 2.55 s (3H, CH<sub>3</sub>CO), 7.0–7.75 m (16H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 156.68 (C<sup>2</sup>); 147.72 (C<sup>5</sup>); 189.17 (CH<sub>3</sub>CO); 24.86 (CH<sub>3</sub>CO); 125.87, 128.86, 130.36, 131.61, 131.77 (NC<sub>6</sub>H<sub>4</sub>Cl-*m*); 117.80, 125.24, 128.39, 134.94, 147.15 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 720 (15) [M]<sup>+</sup>, 360 (10) [C<sub>16</sub>H<sub>11</sub>CIN<sub>3</sub>OS<sub>2</sub>]<sup>+</sup>, 328 (10) [C<sub>16</sub>H<sub>11</sub>CIN<sub>3</sub>OS]<sup>+</sup>, 256 (100) [C<sub>13</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 224 (90) [C<sub>8</sub>H<sub>4</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 167 (25) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (80) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 139 (25) [C<sub>6</sub>H<sub>4</sub>CIN<sub>2</sub>]<sup>+</sup>, 108 (20) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (25) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>. Found, %: C 52.97; H 3.00; N 11.41; S 17.98. C<sub>32</sub>H<sub>22</sub>Cl<sub>2</sub>N<sub>6</sub>O<sub>2</sub>S<sub>4</sub>. Calculated, %: C 53.21; H 3.05; N 11.64; S 17.74.

**Bis[2-(3-*p*-bromophenyl-5-methoxycarbonyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (IIIId).** Yield 46%, mp 196–197°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.9 s (6H, CH<sub>3</sub>O), 7.15–7.9 m (16H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 155.94 (C<sup>2</sup>); 148.09 (C<sup>5</sup>); 157.91 (CH<sub>3</sub>OCO); 53.58 (CH<sub>3</sub>O); 120.16, 124.99, 131.86, 137.34 (NC<sub>6</sub>H<sub>4</sub>Br-*p*); 118.13, 125.39, 127.17, 128.52, 128.74, 139.14 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 842 (10) [M]<sup>+</sup>, 421 (18) [C<sub>16</sub>H<sub>11</sub>BrN<sub>3</sub>O<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 390 (10) [C<sub>16</sub>H<sub>11</sub>BrN<sub>3</sub>O<sub>2</sub>S]<sup>+</sup>, 256 (100) [C<sub>13</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 224 (50) [C<sub>8</sub>H<sub>4</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 184 (10) [C<sub>6</sub>H<sub>4</sub>BrN<sub>2</sub>]<sup>+</sup>, 167 (15) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (80) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 108 (20) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (45) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>. Found, %: C 45.63; H 2.47; N 9.87; S 14.70. C<sub>32</sub>H<sub>22</sub>Br<sub>2</sub>N<sub>6</sub>O<sub>4</sub>S<sub>4</sub>. Calculated, %: C 45.61; H 2.63; N 9.97; S 15.22.

**Bis[2-(5-methoxycarbonyl-3-*p*-methoxyphenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (IIIe).** Yield 84%, mp 75–76°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm: 3.89 s (6H, CH<sub>3</sub>O), 7.1–7.8 m (16H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum,  $\delta_C$ , ppm: 156.37 (C<sup>2</sup>); 148.22 (C<sup>5</sup>); 158.05 (CH<sub>3</sub>OCO); 53.44 (CH<sub>3</sub>O); 114.01, 125.39, 128.85, 158.56 (NC<sub>6</sub>H<sub>4</sub>OCH<sub>3</sub>-*p*); 118.01, 125.13, 126.71, 128.28, 130.87, 137.89 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 744 (10) [M]<sup>+</sup>, 372 (10) [C<sub>17</sub>H<sub>14</sub>N<sub>3</sub>O<sub>3</sub>S<sub>2</sub>]<sup>+</sup>, 340 (20) [C<sub>17</sub>H<sub>14</sub>N<sub>3</sub>O<sub>3</sub>S]<sup>+</sup>, 256 (100) [C<sub>13</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 224 (25) [C<sub>8</sub>H<sub>4</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 167 (25) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (20) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 135 (35) [C<sub>7</sub>H<sub>7</sub>N<sub>2</sub>O]<sup>+</sup>, 108 (35) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (35) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>. Found, %: C 54.73; H 3.85; N 11.71; S 17.97. C<sub>34</sub>H<sub>28</sub>N<sub>6</sub>O<sub>6</sub>S<sub>4</sub>. Calculated, %: C 54.82; H 3.79; N 11.30; S 17.22.

**Bis[2-(5-ethoxycarbonyl-3-phenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (III<sub>f</sub>).** Yield 71%, mp 124–125°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.29 t (6H, CH<sub>3</sub>CH<sub>2</sub>), 4.36 q (4H, CH<sub>3</sub>CH<sub>2</sub>O), 7.1–7.9 m (18H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum, δ, ppm: 156.38 (C<sup>2</sup>); 148.26 (C<sup>5</sup>); 157.49 (CH<sub>3</sub>CH<sub>2</sub>OCO); 62.79 (CH<sub>3</sub>CH<sub>2</sub>O); 13.80 (CH<sub>3</sub>CH<sub>2</sub>); 123.59, 127.84, 128.93, 138.02 (NC<sub>6</sub>H<sub>5</sub>); 118.16, 125.24, 126.87, 128.41, 128.77, 138.84 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 712 (30) [*M*]<sup>+</sup>, 356 (45) [C<sub>17</sub>H<sub>14</sub>N<sub>3</sub>O<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 324 (30) [C<sub>17</sub>H<sub>14</sub>N<sub>3</sub>O<sub>2</sub>S]<sup>+</sup>, 257 (50) [C<sub>13</sub>H<sub>9</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 225 (5) [C<sub>8</sub>H<sub>5</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 167 (25) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (100) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 108 (20) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 105 (35) [C<sub>6</sub>H<sub>5</sub>N<sub>2</sub>]<sup>+</sup>, 91 (70) [C<sub>6</sub>H<sub>5</sub>N]<sup>+</sup>. Found, %: C 45.63; H 2.47; N 9.87; S 14.59. C<sub>34</sub>H<sub>28</sub>N<sub>6</sub>O<sub>4</sub>S<sub>4</sub>. Calculated, %: C 45.61; H 2.63; N 9.97; S 15.22.

**Bis[2-(3-*m*-chlorophenyl-5-ethoxycarbonyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (III<sub>g</sub>).** Yield 67%, mp 113–114°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.35 t (6H, CH<sub>3</sub>CH<sub>2</sub>), 4.40 q (4H, CH<sub>3</sub>CH<sub>2</sub>O), 7.1–8.15 m (16H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 155.83 (C<sup>2</sup>); 147.83 (C<sup>5</sup>); 157.36 (CH<sub>3</sub>CH<sub>2</sub>OCO); 62.94 (CH<sub>3</sub>CH<sub>2</sub>O); 13.79 (CH<sub>3</sub>CH<sub>2</sub>); 121.63, 122.83, 127.45, 128.92, 130.59, 139.13 (NC<sub>6</sub>H<sub>4</sub>Cl-*m*); 118.03, 125.41, 127.14, 128.52, 133.07, 139.51 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 781 (10) [*M*]<sup>+</sup>, 390 (40) [C<sub>17</sub>H<sub>13</sub>ClN<sub>3</sub>O<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 358 (10) [C<sub>17</sub>H<sub>13</sub>ClN<sub>3</sub>O<sub>2</sub>S]<sup>+</sup>, 256 (50) [C<sub>13</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 224 (45) [C<sub>8</sub>H<sub>4</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 167 (40) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (100) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 139 (20) [C<sub>6</sub>H<sub>4</sub>ClN<sub>2</sub>]<sup>+</sup>, 108 (20) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (25) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>. Found, %: C 52.18; H 3.88; N 10.34; S 16.75. C<sub>34</sub>H<sub>26</sub>Cl<sub>2</sub>N<sub>6</sub>O<sub>4</sub>S<sub>4</sub>. Calculated, %: C 52.19; H 3.33; N 10.75; S 16.37.

**Bis[2-(5-ethoxycarbonyl-3-*o*-nitrophenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (III<sub>h</sub>).** Yield 2%, mp 173–174°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.34 t (6H, CH<sub>3</sub>CH<sub>2</sub>); 4.37 q (4H, CH<sub>3</sub>CH<sub>2</sub>O), 7.35–8.3 m (16H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 159.22 (C<sup>2</sup>); 151.02 (C<sup>5</sup>); 165.00 (CH<sub>3</sub>CH<sub>2</sub>OCO); 63.06 (CH<sub>3</sub>CH<sub>2</sub>O); 13.81 (CH<sub>3</sub>CH<sub>2</sub>); 119.14, 121.07, 121.61, 128.11, 129.02, 134.19 (NC<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>-*o*); 119.07, 123.72, 126.29, 128.25, 131.92, 141.22 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 402 (5) [C<sub>17</sub>H<sub>13</sub>N<sub>4</sub>O<sub>4</sub>S<sub>2</sub>]<sup>+</sup>, 369 (5) [C<sub>17</sub>H<sub>13</sub>N<sub>4</sub>O<sub>4</sub>S]<sup>+</sup>, 256 (75) [C<sub>13</sub>H<sub>8</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 224 (100) [C<sub>8</sub>H<sub>4</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 167 (20) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (10) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 122 (40) [C<sub>6</sub>H<sub>4</sub>NO<sub>2</sub>]<sup>+</sup>, 108 (40) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (30) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>.

**Bis[2-(3,5-diphenyl-1,3,4-thiadiazol-2-ylideneamino)phenyl] disulfide (III<sub>j</sub>).** Yield 61%, mp 189–190°C. <sup>1</sup>H NMR spectrum, δ, ppm: 7.1–8.15 m (28H, H<sub>arom</sub>). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 156.33 (C<sup>2</sup>);

149.31 (C<sup>5</sup>); 122.86, 128.91, 130.53 (5-C<sub>6</sub>H<sub>5</sub>); 127.29, 126.17, 128.77, 139.48 (NC<sub>6</sub>H<sub>5</sub>); 118.42, 124.78, 126.47, 126.64, 130.37, 147.02 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 720 (10) [*M*]<sup>+</sup>, 360 (60) [C<sub>20</sub>H<sub>14</sub>N<sub>3</sub>S<sub>2</sub>]<sup>+</sup>, 328 (10) [C<sub>20</sub>H<sub>14</sub>N<sub>3</sub>S]<sup>+</sup>, 257 (30) [C<sub>13</sub>H<sub>9</sub>N<sub>2</sub>S<sub>2</sub>]<sup>+</sup>, 224 (40) [C<sub>8</sub>H<sub>4</sub>N<sub>2</sub>S<sub>3</sub>]<sup>+</sup>, 194 (65) [C<sub>13</sub>H<sub>10</sub>N<sub>2</sub>]<sup>+</sup>, 167 (25) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 154 (80) [C<sub>6</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 108 (5) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 105 (35) [C<sub>6</sub>H<sub>5</sub>N<sub>2</sub>]<sup>+</sup>, 91 (100) [C<sub>6</sub>H<sub>5</sub>N]<sup>+</sup>. Found, %: C 66.80; H 3.85; N 11.27; S 18.15. C<sub>40</sub>H<sub>28</sub>N<sub>6</sub>S<sub>4</sub>. Calculated, %: C 66.58; H 3.88; N 11.65; S 17.75.

**1,3-Benzothiazol-2-yl 2-oxo-*N*-(*m*-chlorophenyl)propanehydrazone thioate (IV<sub>b</sub>).** Yield 38%, mp 191–192°C. <sup>1</sup>H NMR spectrum, δ, ppm: 2.6 s (3H, CH<sub>3</sub>CO, *Z* isomer), 2.65 s (3H, CH<sub>3</sub>CO, *E* isomer), 7.03–7.84 m (16H, H<sub>arom</sub>, *E* and *Z* isomers), 11.4 s (1H, NH, *Z* isomer), 13.41 s (1H, NH, *E* isomer). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 128.22 (C<sup>1</sup>); 163.26 (C<sup>2</sup>); 191.61 (CH<sub>3</sub>CO); 25.57 (CH<sub>3</sub>CO); 114.06, 115.16, 121.29, 131.00, 133.80, 143.69 (NC<sub>6</sub>H<sub>4</sub>Cl-*m*); 121.71, 123.09, 124.61, 126.37, 134.87, 153.06 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 361 (15) [*M*]<sup>+</sup>, 167 (35) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 134 (5) [C<sub>7</sub>H<sub>4</sub>NS]<sup>+</sup>, 122 (10) [C<sub>6</sub>H<sub>4</sub>NS]<sup>+</sup>, 108 (15) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (15) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>. Found, %: C 53.24; H 3.28; N 10.28; S 17.19. C<sub>16</sub>H<sub>12</sub>ClN<sub>3</sub>OS<sub>2</sub>. Calculated, %: C 53.06; H 3.32; N 9.95; S 17.69.

**Methyl (1,3-benzothiazol-2-ylsulfanyl)(*p*-bromophenylhydrazone)acetate (IV<sub>d</sub>).** Yield 35%, mp 154–155°C. <sup>1</sup>H NMR spectrum, δ, ppm: 3.8 s (3H, CH<sub>3</sub>O, *Z* isomer), 3.95 s (3H, CH<sub>3</sub>O, *E* isomer), 7.2–7.9 m (16H, H<sub>arom</sub>, *E* and *Z* isomers), 11.3 s (1H, NH, *Z* isomer), 13.7 s (1H, NH, *E* isomer). <sup>13</sup>C NMR spectrum, δ<sub>C</sub>, ppm: 130.70 (C<sup>1</sup>); 164.62 (C<sup>2</sup>); 163.88 (CH<sub>3</sub>OCO); 53.36 (CH<sub>3</sub>O); 115.12, 117.33, 131.01, 132.00, 140.68 (NC<sub>6</sub>H<sub>4</sub>Br-*p*); 120.14, 123.04, 124.67, 126.32, 134.86, 152.37 (C<sub>6</sub>H<sub>4</sub>). Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 423 (10) [*M*]<sup>+</sup>, 256 (10) [C<sub>9</sub>H<sub>8</sub>N<sub>2</sub>O<sub>2</sub>Br]<sup>+</sup>, 167 (80) [C<sub>7</sub>H<sub>4</sub>NS<sub>2</sub>]<sup>+</sup>, 134 (10) [C<sub>7</sub>H<sub>4</sub>NS]<sup>+</sup>, 122 (25) [C<sub>6</sub>H<sub>4</sub>NS]<sup>+</sup>, 108 (40) [C<sub>6</sub>H<sub>4</sub>S]<sup>+</sup>, 90 (50) [C<sub>6</sub>H<sub>4</sub>N]<sup>+</sup>. Found, %: C 45.94; H 3.22; N 9.77; S 15.49. C<sub>16</sub>H<sub>12</sub>BrN<sub>3</sub>O<sub>2</sub>S<sub>2</sub>. Calculated, %: C 45.46; H 2.85; N 9.94; S 15.15.

**Ethyl (1,3-benzothiazol-2-ylsulfanyl)(phenylhydrazono)acetate (IV<sub>f</sub>).** Yield 0.05%.

**Ethyl (1,3-benzothiazol-2-ylsulfanyl)(*m*-chlorophenylhydrazone)acetate (IV<sub>g</sub>).** Yield 26%, mp 169–170°C. <sup>1</sup>H NMR spectrum, δ, ppm: 1.34 t (3H, CH<sub>3</sub>CH<sub>2</sub>, *Z* isomer), 1.44 t (3H, CH<sub>3</sub>CH<sub>2</sub>, *E* isomer), 4.29 q (2H, CH<sub>3</sub>CH<sub>2</sub>O, *Z* isomer), 4.41 q (2H, CH<sub>3</sub>CH<sub>2</sub>O, *E* isomer), 6.98–7.91 m [16H, H<sub>arom</sub>, *E* and *Z* isomers], 11.3 s (1H, NH, *Z* isomer), 13.65 s (1H,



NH, *E* isomer).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: 130.09 ( $\text{C}^1$ ); 162.83 ( $\text{C}^2$ ); 160.75 ( $\text{CH}_3\text{CH}_2\text{OCO}$ ); 61.99 ( $\text{CH}_3\text{CH}_2\text{O}$ ); 14.06 ( $\text{CH}_3\text{CH}_2$ ); 113.76, 115.27, 121.29, 129.91, 133.47, 143.78 ( $\text{NC}_6\text{H}_4\text{Cl-}m$ ); 121.17, 122.48, 124.28, 125.83, 135.01, 153.18 ( $\text{C}_6\text{H}_4$ ). Found, %: C 52.18; H 3.88; N 10.70; S 16.75.  $\text{C}_{17}\text{H}_{14}\text{ClN}_3\text{O}_2\text{S}_2$ . Calculated, %: C 52.10; H 3.60; N 10.72; S 16.36.

**Ethyl (1,3-benzothiazol-2-ylsulfanyl)(*o*-nitrophenylhydrazono)acetate (IVh).** Yield 55% (*E* isomer), 30% (*Z* isomer); mp 167–168°C (*E* isomer), 78–79°C (*Z* isomer).  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.12 t (3H,  $\text{CH}_3\text{CH}_2$ , *E* isomer), 1.26 t (3H,  $\text{CH}_3\text{CH}_2$ , *Z* isomer), 4.30 q (2H,  $\text{CH}_3\text{CH}_2\text{O}$ , *E* isomer), 4.08 q (2H,  $\text{CH}_3\text{CH}_2\text{O}$ , *Z* isomer), 7.13–8.15 m (8H,  $\text{H}_{\text{arom}}$ , *Z* isomer), 7.23–8.24 m (8H,  $\text{H}_{\text{arom}}$ , *E* isomer), 12.9 s (1H, NH, *Z* isomer), 14.05 s (1H, NH, *E* isomer).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: *E* isomer: 126.56 ( $\text{C}^1$ ); 166.63 ( $\text{C}^2$ ); 160.38 ( $\text{CH}_3\text{CH}_2\text{OCO}$ ); 62.59 ( $\text{CH}_3\text{CH}_2\text{O}$ ); 13.58 ( $\text{CH}_3\text{CH}_2$ ); 115.86, 124.98, 125.87, 135.13, 136.56, 138.14 ( $\text{NC}_6\text{H}_4\text{NO}_2\text{-}o$ ); 121.66, 122.63, 125.87, 126.46, 133.57, 152.66 ( $\text{C}_6\text{H}_4$ ). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 402 (50) [ $M$ ] $^+$ , 167 (80) [ $\text{C}_7\text{H}_4\text{NS}_2$ ] $^+$ , 134 (30) [ $\text{C}_7\text{H}_4\text{NS}$ ] $^+$ , 122 (20) [ $\text{C}_6\text{H}_4\text{NS}$ ] $^+$ , 108 (30) [ $\text{C}_6\text{H}_4\text{S}$ ] $^+$ , 90 (15) [ $\text{C}_6\text{H}_4\text{N}$ ] $^+$ . Found, %: C 51.02; H 3.80; N 13.33; S 16.32.  $\text{C}_{17}\text{H}_{14}\text{N}_4\text{O}_4\text{S}_2$ . Calculated, %: C 50.69; H 3.48; N 13.91; S 15.90.

**Ethyl (1,3-benzothiazol-2-ylsulfanyl)(*o*-chlorophenylhydrazono)acetate (IVi).** Yield 80% (*E* isomer), mp 106–107°C; traces of the *Z* isomer were also detected.  $^1\text{H}$  NMR spectrum,  $\delta$ , ppm: 1.15 t (3H,  $\text{CH}_3\text{CH}_2$ , *E* isomer), 1.36 t (3H,  $\text{CH}_3\text{CH}_2$ , *Z* isomer), 4.30 q (2H,  $\text{CH}_3\text{CH}_2\text{O}$ , *E* isomer), 4.32 q (2H,  $\text{CH}_3\text{CH}_2\text{O}$ , *Z* isomer), 7.12–7.84 m (8H,  $\text{H}_{\text{arom}}$ , *E* isomer), 7.02–7.85 m (8H,  $\text{H}_{\text{arom}}$ , *Z* isomer), 9.85 s (1H, NH, *Z* isomer), 12.95 s (1H, NH, *E* isomer).  $^{13}\text{C}$  NMR spectrum,  $\delta_{\text{C}}$ , ppm: *E* isomer: 128.97 ( $\text{C}^1$ ); 166.11 ( $\text{C}^2$ ); 161.96 ( $\text{CH}_3\text{CH}_2\text{OCO}$ ); 62.29 ( $\text{CH}_3\text{CH}_2\text{O}$ ); 13.61 ( $\text{CH}_3\text{CH}_2$ ); 114.84, 118.83, 126.45, 128.65, 129.65, 137.74 ( $\text{NC}_6\text{H}_4\text{Cl-}o$ ); 121.46, 121.83, 124.58, 124.68, 134.86, 152.90 ( $\text{C}_6\text{H}_4$ ). Mass spectrum,  $m/z$  ( $I_{\text{rel}}$ , %): 391 (20) [ $M$ ] $^+$ , 225 (15) [ $\text{C}_{10}\text{H}_{11}\text{ClN}_2\text{O}_2$ ] $^+$ , 167

(60) [ $\text{C}_7\text{H}_4\text{NS}_2$ ] $^+$ , 134 (10) [ $\text{C}_7\text{H}_4\text{NS}$ ] $^+$ , 122 (15) [ $\text{C}_6\text{H}_4\text{NS}$ ] $^+$ , 108 (35) [ $\text{C}_6\text{H}_4\text{S}$ ] $^+$ , 90 (35) [ $\text{C}_6\text{H}_4\text{N}$ ] $^+$ . Found, %: C 53.12; H 3.63; N 10.66; S 16.74.  $\text{C}_{17}\text{H}_{14}\text{ClN}_3\text{O}_2\text{S}_2$ . Calculated, %: C 52.05; H 3.57; N 10.71; S 16.33.

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## REFERENCES

- Labeish, N.N., Oparin, D.A., Bel'skii, V.K., and Galishev, V.A., *Russ. J. Org. Chem.*, 1997, vol. 33, p. 381.
- Huisgen, R., Grashey, R., Seidel, M., Knupfer, H., and Schmidt, R., *Justus Liebigs Ann. Chem.*, 1962, vol. 658, p. 169.
- Al'fonsov, V.A., Belen'kii, L.I., *et al.*, *Poluchenie i svoistva organicheskikh soedinenii sery* (Synthesis and Properties of Organic Sulfur Compounds), Belen'kii, L.I., Ed., Moscow: Khimiya, 1998, p. 81.
- Wolkoff, P., Nemeth, C.S., and Gibson, M.S., *Can. J. Chem.*, 1975, vol. 53, p. 3211.
- Abdelhamid, A.O. and Attaby, F.A., *J. Heterocycl. Chem.*, 1976, vol. 13, p. 45.
- Wolkoff, P. and Hammerum, S., *Acta Chem. Scand., Ser. B*, 1976, vol. 30, p. 831.
- Abdelhamid, A.O. and Attaby, F.A., *J. Heterocycl. Chem.*, 1991, vol. 28, p. 41.
- Abdallah Magda, A., Mosselhi Mosselhi, A.N., Abbas Ikhlass, M., Fahmi Abdel-Gawad, A., and Shawali, Ahmed S., *J. Chem. Res., Synop.*, 1995, no. 9, p. 370.
- Stavrovskaya, V.I. and Kolosova, M.O., *Zh. Obshch. Khim.*, 1960, vol. 30, p. 689.
- Chebonnet, A., *Theses*, Guillemonat, M.A., Metzger, J., and Julg, A., Eds., 1962, p. 62.
- Safaev, A.S. and Kadyrov, A.K., *Sintez i prevraschenie soedinenii ryada piperidina, tiazola i imidazolina* (Synthesis and Transformations of Compounds of the Piperidine, Thiazole, and Dihydroimidazole Series), Tashkent: Fan, 1984, chap. IV.
- Halasa, A.F. and Smith, Jr., *J. Org. Chem.*, 1971, vol. 36, p. 636.
- Huisgen, R., Seidel, M., Wallbilich, G., and Knupfer, H., *Tetrahedron*, 1962, vol. 17, p. 3.